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In chemistry. Standards are used in analytical chemistry. Here, a primary standard is typically a reagent which can be weighed easily, and that its weight is truly representative of the number of moles of substance contained. Features of a primary standard include: High purity, stability, and reactivity)

[Primary standard - Wikipedia](#)

High School Chemistry Help » Acid-Base Chemistry » Acids and Bases » Calculating pH and pOH Example Question #1 : Calculating P H And Sodium hydroxide is a strong base.

[Calculating pH and pOH - High School Chemistry](#)

In chemistry and biochemistry, the Henderson–Hasselbalch equation = + ? ([] []) can be used to estimate the pH of a buffer solution. The of the acid dissociation constant, K_a , of the acid is known or assumed. The pH is calculated for given values of the concentrations of the salt, MA, of its conjugate base, A^- ; for example, the solution may contain ...

[Henderson–Hasselbalch equation - Wikipedia](#)

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